13. The following statement represents a nuclear reaction.

$${}^{12}_{6}\text{C} + {}^{4}_{2}\text{He} \rightarrow {}^{16}_{8}\text{O} + \gamma$$

The total mass of the particles before the reaction is 26.572×10^{-27} kg.

The total mass of the particles after the reaction is $26.560\times 10^{-27}\ kg.$

The energy released in this reaction is

A
$$1.20 \times 10^{-29} \text{ J}$$

.....

B $3.60 \times 10^{-21} \text{ J}$

C $5.40 \times 10^{-13} \text{ J}$

D $1.08 \times 10^{-12} \, \text{J}$

E $2.39 \times 10^{-9} \text{ J.}$